# **Platform Chemicals from Biomass Fermentation**

## a Report Prepared for Chemistry Innovation KTN

by

Miller-Klein Associates Ltd

Saith Ffynnnon Farm

Whitford

Flintshire



### **Platform Chemicals from Biomass Fermentation**

The following table shows the main platform chemicals and derivatives that could be derived from processing of plant biomass in a biorefinery, and includes glycerol generated as a by-product from bio-diesel.

The material name, structure, applications and current production volumes are given if available.

The principal sources of information used were:

- "Medium and Long-term Opportunities and Risks of the Biotechnological Production of Bulk Chemicals from Renewable Resources -The Potential of White Biotechnology", final report of the BREW project, University of Utrecht, 2006
- "Top Value Added Chemicals from Biomass: Volume I—Results of Screening for Potential Candidates from Sugars and Synthesis Gas" National Renewable Energy Laboratory, Pacific NorthWest National Laboratory, US Department of Energy Biomass Programme, 2004
- "Handbook of Industrial Chemistry Organic Chemicals" Ali, Mohammad Farhat; El Ali, Bassam M.; Speight, James G. © 2005 McGraw-Hill
- www.wikipedia.org

The table covers a selection of the potential platform chemicals and derivatives that are reported in these and other publications. The criteria for selection were:

- Could be produced from a feedstock suitable for biofuel production
- Known biotechnological or chemical route to material
- Known or predicted industrial application

Direct extracts from plant biomass (eg plant sterols, artemisinin, plant waxes etc) have been excluded from this table.

#### **General Points**

The chemicals produced from a fermentation biorefinery have two important characteristics:

They are typically oxygenates

Starting from sugars and starches which are highly oxygenated, the vast majority of potential platform chemicals that have been described also carry at least one oxygen atom – sometimes several. This is important for downstream processing. The petrochemical based chemical industry traditionally manipulates the main structure of a molecule as a hydrocarbon; adding heteroatoms at a later stage in the process. In contrast, when starting from bio-based platform chemicals oxygen is almost always already present. This dramatically changes the synthetic strategies for reaching target molecules. As a result bio-based platform chemicals cannot easily be introduced into the material flows for the current mainstream chemical industry.

They are typically multifunctional

The majority of the platform chemicals and their derivatives have more than one functional group available for further reaction. This multifunctional nature makes them extremely flexible when devising synthetic strategies, and makes them particularly appropriate for polymerisation reactions; e.g. the production of polyesters, polyethers and polyurethanes.

Given these characteristics, the platform chemicals and their derivatives listed currently find application in:

- Fuel and fuel additives
- Solvents
- Polymers
- Coatings adhesives sealants and elastomers
- Lubricants and hydraulic fluids
- Surfactants
- Cosmetic ingredients

#### Notes on the Tables

The tables in this appendix show information about the platform chemicals and their derivatives. The column headings are:

<ul> <li>Carbon number</li> </ul>	Number of carbon atoms in the backbone of the platform chemical. Ranges from 2 to 6.
<ul> <li>Platform chemical</li> </ul>	The name of the platform chemical with a specific carbon number.
<ul> <li>Derivative</li> </ul>	The name of a derivative of a specific platform chemical.
<ul> <li>Structure</li> </ul>	The molecular structure of the platform chemical or derivative. In cases where the structure is variable, either no structure or a 'typical' structure is given.
<ul> <li>Applications</li> </ul>	Known or predicted applications of the platform chemical or derivative. Almost all of these are current commercial applications.
<ul> <li>Production</li> </ul>	Information on the global production of the chemical where known.

Carbon Number	Platform Chemical	Derivatives	Structure	Applications	Production
2	Ethanol		ОН	<ul><li>70% fuel,</li><li>20% industrial,</li><li>10% beverages</li></ul>	40bn litres in 2003. >90% by fermentation.
		Ethylene	H <sub>2</sub> C=CH <sub>2</sub>	Key chemical building block : Polyethylene PVC Ethylene oxide Ethylene glycols	94 Mt (2002)
		Ethyl <i>tert</i> -butyl ether	СH <sub>3</sub> H <sub>3</sub> C — СH <sub>3</sub>	Fuel additive. Competes with MTBE and Ethanol	5 bn litres p.a.
		Ethyl esters		Greener solvents	1 Mt p.a. of ethyl acetate
		Ethylethers	R <sup>O</sup> CH <sub>3</sub>	Solvents and medical applications	
		Glycolethers	R <sub>0</sub> OH	Solvents – cellosolve and carbitols	
		Ethylamine	NH <sub>2</sub> H <sub>3</sub> C	<ul><li>Solvent</li><li>Synthetic intermediate</li></ul>	80 kt p.a.
		Acetaldehyde	н₃сО	<ul><li>Production of:</li><li>Acetic acid and anhydride</li><li>Pyridine</li></ul>	1.35 Mt (1993)
2	Acetic acid		H₃C → OH	<ul><li>Production of:</li><li>Vinyl acetate</li><li>Acetic anhydride</li><li>Used as a solvent</li></ul>	>7 Mt p.a. 190 kt by fermentation

Acetic anhydride	CH <sub>3</sub> CH <sub>3</sub>	Acetylating agent for alcohols and amines Dehydrating agent	1-2 Mt p.a.
Vinyl acetate	н₂с=_0сн₃	Monomer for polymer and copolymer production: Polyvinylacetate Polyvinylalcohol Ethylene – vinyl acetate	4 Mt p.a.(2002)
Ethyl acetate	о н <sub>з</sub> с— сн <sub>з</sub>	Green solvent	1 Mt p.a.
Chloroacetic acid	CI OH	<ul> <li>Production of:</li> <li>Carboxymethylcellulose</li> <li>Phenoxy herbicides</li> <li>As a difunctional reactive intermediate</li> </ul>	370 kt p.a.
Peracetic acid	HO <sup>O</sup> CH <sub>3</sub>	<ul> <li>Antimicrobial</li> <li>Bleaching agent</li> <li>Oxidising agent to produce epoxides and alcohols</li> </ul>	18 kt p.a.

Carbon Number	Platform Chemical	Derivatives	Structure	Applications	Production
3	Lactic acid		HO O H <sub>3</sub> C OH	<ul><li>Food and beverages</li><li>Green solvents</li><li>Bio-polymers</li></ul>	150 kt in 2002 – almost entirely by fermentation
		Polylactic acid		<ul> <li>Compostable polymer from renewable source</li> <li>Packaging</li> <li>Fibres and Textiles</li> </ul>	140 kt p.a. production capacity
		Lactic acid salts	Ŭ	<ul> <li>Food and beverages</li> <li>Permeation enhancer (medical)</li> </ul>	
		Esters		<ul> <li>Green solvents (eg ethyl lactate)</li> </ul>	5 kt p.a.
		Chiral esters		<ul> <li>Chiral synthons for pharmaceuticals and agrochemicals</li> </ul>	
		Acetaldehyde	н₃сО	Common 2 carbon building block in synthesis	
		Acrylic acid		<ul> <li>Acrylates</li> <li>Polymers and copolymers for coatings, adhesives, sealants and elastomers</li> </ul>	>2 Mt p.a.
		Lactamides	R HO NH R H <sub>3</sub> C O	Plasticisers	
		1,2-Propanediol	но снз	<ul> <li>moisturizer in medicines, cosmetics, food, and tobacco products</li> <li>medical and sexual lubricant</li> <li>emulsification agent</li> <li>solvent for food colors and flavourings</li> </ul>	0.9 Mt p.a. €1000-€1200/t

				<ul> <li>humectant</li> <li>carrier in fragrance oils</li> <li>non-toxic antifreeze</li> <li>main ingredient in cosmetic products, including baby wipes, bubble baths, deodorants, shampoos, and hair dyes</li> <li>working fluid in hydraulic presses</li> </ul>	
		Propionic acid	H0 H <sub>3</sub> C	<ul> <li>Preservative</li> <li>Esters used as solvents</li> <li>Chemical intermediate</li> </ul>	130 kt (1989)
		2,3-Pentadione	H <sub>3</sub> C CH <sub>3</sub>	<ul> <li>Solvent</li> <li>Flavour synthesis</li> <li>Chemical intermediate</li> </ul>	
		Oxalic acid	но о	<ul> <li>Mordant for dyeing</li> <li>Household chemicals</li> <li>Rust remover and rust proofer</li> </ul>	
		Pyruvic acid	о о Н <sub>3</sub> С ОН	<ul> <li>Medical applications</li> <li>Production of plant growth regulators</li> </ul>	
3	3-Hydroxypropionic acid		он	<ul><li>Acrylic polymers</li><li>Speciality polyesters</li></ul>	
		1,3-Propanediol	HOOH	<ul> <li>Polymer building block (PTT)</li> </ul>	80 kt p.a.
		Malonic acid	но он	<ul> <li>Synthetic intermediate for pharmaceuticals, agrochemicals and flavors &amp; fragrances compounds.</li> </ul>	

		Acrylic acid	HO	<ul> <li>Acrylates</li> </ul>	>2 Mt p.a.
		Aci yile aciu		<ul> <li>Activities</li> <li>Polymers and copolymers for</li> </ul>	~2 ivit p.a.
			0—(	coatings, adhesives, sealants and	
			н,с=/	elastomers	
		Acritamida	-		
		Acrylamide	H <sub>2</sub> N	Production of polymers and modified	
			)—o	<ul><li>copolymers for:</li><li>waste and sewage treatment,</li></ul>	
				<ul> <li>waste and sewage treatment,</li> <li>paper and pulp manufacturing</li> </ul>	
			H <sub>2</sub> C=	<ul> <li>paper and pup manufacturing</li> <li>oil recovery and ore processes</li> </ul>	
				<ul> <li>on recovery and one processes</li> <li>soil stabilizer</li> </ul>	
				<ul> <li>adhesive coating</li> </ul>	
				<ul> <li>food processing.</li> </ul>	
		Methyl acrylate	0	Production of:	
			Ĭ	<ul> <li>coatings, elastomers, adhesives,</li> </ul>	
			H <sub>2</sub> C	thickeners.	
			- ··· · · · · · · · · · · · · · · · · ·	<ul> <li>amphoteric surfactants,</li> </ul>	
				<ul> <li>fibers, plastics, textiles and inks</li> </ul>	
			CH3	Chemical synthesis.	
3	1, 3-Propanediol		HOOH	<ul> <li>Polymer building block (PTT)</li> </ul>	80 kt p.a.
		Polytrimethylene	~ ~ ~	<ul> <li>Fibres and textiles</li> </ul>	>100 kt p.a. installed capacity
		terephthalate		<ul> <li>Engineering plastics</li> </ul>	, ,
				Brand names Sorona and Corterra	
		Polyurethanes		<ul> <li>Chain extender for thermoplastic</li> </ul>	
				polyurethanes – replacement for	
				1,4-butanediol	
		Copolyester ethers		<ul> <li>High performance elastomers</li> </ul>	
3	Glycerol		ŎН	<ul> <li>Triacetin</li> </ul>	600 kt p.a. in Europe (2006).
				<ul> <li>Food additive</li> </ul>	Increased from ~200 kt in
				<ul> <li>Pharmaceuticals</li> </ul>	2000. Expected to continue to
				<ul> <li>Personal Care</li> </ul>	rise due to increased
			он он	<ul> <li>Polyols</li> </ul>	production of biodiesel
				<ul> <li>Alkyd resins</li> </ul>	
				Tobacco	2006 price ~500€/t – a
				<ul> <li>Explosives</li> </ul>	historically low level.
				Detergents	
				Cellophane	4.0.14
		Epichlorohydrin	o. CI	Epoxy resins	1.2 Mt p.a.
1			$\sim$ /	<ul> <li>Paper chemicals</li> </ul>	\$2300/t
					φ2000/1

		Water treatment	
		<ul> <li>Polyglycerols</li> </ul>	
1,2-propanediol	но СН3	<ul> <li>moisturizer in medicines, cosmetics, food, and tobacco products</li> <li>medical and sexual lubricant</li> <li>emulsification agent</li> <li>solvent for food colors and flavourings</li> <li>humectant</li> <li>carrier in fragrance oils</li> <li>non-toxic antifreeze</li> <li>main ingredient in cosmetic products, including baby wipes, bubble baths, deodorants, shampoos, and hair dyes</li> <li>working fluid in hydraulic presses</li> </ul>	0.9 Mt p.a. €1000-€1200/t
1,3-Propanediol	HOOH	<ul> <li>Polymer building block (PTT)</li> </ul>	80 kt p.a.
Mono-, di- and tri-esters		<ul> <li>Food additives</li> <li>Emulsifiers</li> </ul>	
Polyglycerols		<ul> <li>Non-ionic surfactants.</li> <li>Emulsifiers in food, cosmetics etc</li> <li>Antifogging agents in polyolefin films</li> </ul>	
Glyceric acid	ОН ОН ОН	<ul> <li>Potential for PLA analogues with different properties</li> </ul>	
Acrolein	0=/ <sup>-CH2</sup>	<ul><li>Acrylic acid esters,</li><li>Polymers,</li><li>Detergents</li></ul>	

Carbon Number	Platform Chemical	Derivatives	Structure	Applications	Production
4 5	Succinic acid		но	<ul> <li>Sweetener in food and beverages</li> </ul>	16 kt p.a.
		1,4-Butanediol	но	<ul> <li>Polybutylene terephthalate (PBT)</li> <li>Polybutylene succinate (PBS)</li> </ul>	512 kt p.a. (1995)
		γ-Butyrolactone		<ul> <li>Solvent for polymers and agrochemicals.</li> <li>Intermediate in the manufacture of pyrrolidone derivatives</li> </ul>	
		Tetrahydrofuran	$\langle \rangle$	<ul> <li>Solvent</li> <li>Thermoplastic polyurethanes,</li> <li>Elastic fibres,</li> <li>Moulded elastomers,</li> <li>Copolyesters and copolyamides.</li> </ul>	140 kt p.a. (1992)
		<i>N</i> -Methyl-2-pyrrolidone (NMP)	O N CH3	<ul> <li>Solvent</li> <li>Reaction medium</li> </ul>	30 kt p.a.
		Di-esters		<ul> <li>Green solvents</li> <li>Fuel oxygenates</li> </ul>	
		Polyamides	2		
		Polyesters		<ul> <li>e.g. polybutylene succinate</li> </ul>	

4	Fumaric Acid		но	<ul> <li>Unsaturated polyester resins</li> <li>Food additive</li> <li>Animal feed</li> <li>Dye mordant</li> <li>Polyhydric alcohols</li> </ul>	12 kt p.a.
		1,4-Butanediol	НО	<ul> <li>Polybutylene terephthalate (PBT)</li> <li>Polybutylene succinate (PBS)</li> </ul>	512 kt p.a. (1995)
		γ-Butyrolactone		<ul> <li>Solvent for polymers and agrochemicals.</li> <li>Intermediate in the manufacture of pyrrolidone derivatives</li> </ul>	
		Tetrahydrofuran	$\bigcirc$	<ul> <li>Solvent</li> <li>Thermoplastic polyurethanes,</li> <li>Elastic fibres,</li> <li>Moulded elastomers,</li> <li>Copolyesters and copolyamides.</li> </ul>	140 kt p.a. (1992)
		Unsaturated polyesters		Replaces maleic acid and maleic anhydride for polyesters with: Improved thermal stability Greater hardness	
		L-Aspartic acid		Production of sweetener aspartame	13 kt p.a.
		L-Alanine			
		Succinic acid	но	<ul> <li>Sweetener in food and beverages</li> </ul>	16 kt p.a.

4	Aspartic Acid			Production of sweetener aspartame	13 kt p.a.
		Amino-γ-butyrolactone		Potential polymer and solvent applications	
		Aspartic anhydride		Potential polymer and solvent applications	
		3-Aminotetrahydrofuran		Potential polymer and solvent applications	
		2-Amino-1,4,-butanediol	HO HO OH	Potential polymer and solvent applications	
		Polyaspartic acid		<ul> <li>Potential substitute for polyacrylates and polycarboxylates. Potential applications in:</li> <li>Detergents,</li> <li>Water treatment,</li> <li>Corrosion inhibition</li> <li>Super-absorbers.</li> </ul>	
4	1-Butanol		Н <sub>3</sub> СОН	<ul> <li>Solvent</li> <li>Thinner for varnishes and lacquers</li> <li>Plasticizers</li> <li>Butylamines</li> <li>Butyl acetate,</li> <li>Acrylic esters,</li> <li>Glycol esters.</li> </ul>	2 Mt p.a.

4	1,4-Butanediol		НО	<ul> <li>Polybutylene terephthalate (PBT)</li> <li>Polybutylene succinate (PBS)</li> </ul>	512 kt p.a. (1995)
		Polybutylene terephthalate (PBT)			340 kt p.a. (1997)
		Polybutylene succinate (PBS)			33 kt p.a. installed capacity 2006
		Polyurethanes			
		γ-Butyrolactone		<ul> <li>Solvent for polymers and agrochemicals.</li> <li>Intermediate in the manufacture of pyrrolidone derivatives</li> </ul>	
		Tetrahydrofuran	$\langle \rangle$	<ul> <li>Solvent</li> <li>Thermoplastic polyurethanes,</li> <li>Elastic fibres,</li> <li>Moulded elastomers,</li> <li>Copolyesters and copolyamides.</li> </ul>	140 kt p.a. (1992)
		Adipic acid	но	<ul> <li>Nylon 66</li> <li>Lubricant esters</li> <li>Plasticizers</li> <li>Polyurethanes</li> </ul>	2.5 Mt p.a.
		Pyrrolidones	R-N	Chemical intermediate	

Carbon Number	Platform Chemical	Derivatives	Structure	Applications	Production
5	Xylose / Arabinose		но он	<ul> <li>Source of C5 sugars for chemical synthesis</li> </ul>	
		1,2,4-Butanetriol	ноон	<ul> <li>Explosive</li> <li>Propellant</li> <li>Chiral synthon</li> </ul>	
		Xylitol / Arabinitol	он он но но он	<ul> <li>Sweetener</li> </ul>	
		Xylaric, Xylonic, Arabonic, Arabinoic acid		<ul> <li>Potential uses in new polymers</li> </ul>	
		Polyesters		<ul> <li>Xylitol/arabinitol with other glycols for unsaturated polyesters</li> </ul>	
		Ethylene and propylene glycol	ноон	<ul><li>Deicer</li><li>Automotive antifreeze</li></ul>	

			но ОН СН3	<ul> <li>Building block unsaturated polyesters</li> </ul>	
		Levulinic acid	н <sub>з</sub> с он	See below	See below
5	Levulinic acid		н <sub>з</sub> с Он	Chemical intermediate	450 t p.a.
		Methyltetrahydrofuran	H <sub>3</sub> C	<ul><li>Solvent</li><li>Fuel oxygenate</li><li>Chemical intermediate</li></ul>	
		Esters		<ul> <li>Fuel oxygenates</li> </ul>	
		γ-Valerolactone	H <sub>3</sub> C 0 0	<ul> <li>Solvent</li> <li>Chemical intermediate</li> </ul>	
		5-Methyl-2-pyrrolidone	H <sub>3</sub> C N O	<ul> <li>Solvent</li> <li>Chemical intermediate</li> </ul>	
		δ-Amino-levulinic acid		Herbicide	
		Diphenolic acid		<ul> <li>Potential replacement for Bisphenol-A</li> </ul>	
		1,4-Pentanediol	HOCH3	<ul> <li>Diol for polyesters</li> </ul>	

		β-Acetyl-acrylic acid	0	Co-polymers	
			Н <sub>3</sub> С ОН		
		Succinic acid	но	<ul> <li>Sweetener in food and beverages</li> </ul>	16 kt p.a.
		Acrylic acid		<ul> <li>Acrylates</li> <li>Polymers and copolymers for coatings, adhesives, sealants and elastomers</li> </ul>	>2 Mt p.a.
5	Furfural			<ul> <li>Extraction solvent</li> <li>Nematicide</li> <li>Fungicide</li> <li>Resins</li> </ul>	200 kt – 300 kt
		Furfuryl alcohol	ОН	Resins	120 kt – 180 kt p.a.
		Furoic acid	ОН		
		Tetrahydrofurfuryl alcohol	О	<ul><li>Solvent</li><li>Industrial cleaning</li></ul>	
		Furfuryl amine	NH <sub>2</sub>	Pharmaceuticals	
		Tetrahydrofuran	$\langle \rangle$	<ul> <li>Solvent</li> <li>Thermoplastic polyurethanes,</li> <li>Elastic fibres,</li> <li>Moulded elastomers,</li> <li>Copolyesters and copolyamides.</li> </ul>	140 kt p.a. (1992)

Levulinic acid	H <sub>3</sub> C OH	Chemical intermediate	450 t p.a.
Maleic anhydride	0	<ul> <li>Important chemical intermediate</li> </ul>	
Thermoset resins		<ul> <li>Foundry binder</li> </ul>	

Carbon Number	Platform Chemical	Derivatives	Structure	Applications	Production
6	2,5- Furandicarboxylic acid		но он	<ul> <li>Potential as monomer to replace terephthalic acid</li> </ul>	
		2,5-Dihydroxymethylfuran	но он	<ul> <li>Potential use in new polyesters</li> </ul>	
		2,5-Dihydroxymethyl- tetrahydrofuran	но он	<ul> <li>Potential use in new polyesters</li> </ul>	
		2,5-Bis(aminomethyl)- tetrahydrofuran	H <sub>2</sub> N NH <sub>2</sub>	<ul> <li>Potential use in new polyamides</li> </ul>	
6	Sorbitol		он он но он он он он	<ul> <li>Sweetener</li> <li>Humectant</li> <li>Thickener</li> <li>Cryo-stabiliser</li> <li>Amateur rocket fuel</li> </ul>	1.1 Mt p.a.
		1,4-Sorbitan	ОН ОН ОН	<ul> <li>Esters used as non-ionic surfactants</li> <li>Solubiliser</li> <li>Emulsifier</li> </ul>	50 kt p.s.
		Isosorbide	но о ОН	<ul> <li>Pharmaceutical (vasodilator)</li> <li>Esters used as solvent in cosmetics</li> <li>Potential for use as diol in polyesters</li> <li>Potential for plasticizers</li> </ul>	800 t p.a.
		Polyetherpolyols		<ul> <li>Polyurethane synthesis</li> </ul>	

		Ascorbic acid	НО ОН ОН	<ul> <li>Salts – antioxidants for aqueous systems</li> <li>Esters – antioxidants for non- aqueous systems</li> </ul>	
6	5- Hydroxymethylfufural		O OH	Phenolic resins	
		5-Hydroxymethyl-furoic acid	но		
		2,5-Furan dicarboxylic acid	но он	<ul> <li>Green alternative to terephthalic acid</li> </ul>	
		Furandialdehyde	o o o	<ul> <li>Potential for new polyesters</li> </ul>	
		2,5-Dihydroxymethylfuran	но он	<ul> <li>Potential for new polyesters</li> </ul>	
		2,5-Diaminomethylfuran	H <sub>2</sub> N NH <sub>2</sub>	<ul> <li>Potential for new polyamides</li> </ul>	
		2,5-Dihydroxymethyl- tetrahydrofuran	но Он	<ul> <li>Potential for new polyesters</li> </ul>	
6	Adipic acid		OH O O O O O O O O H	<ul> <li>Nylon 66</li> <li>Lubricant esters</li> <li>Plasticizers</li> <li>Polyurethanes</li> </ul>	2.5 Mt p.a.

Polyamides		<ul><li>Engineering plastics</li><li>Fibres</li></ul>	
Polyurethanes			
Esters	•	<ul> <li>Lubricants</li> </ul>	
	•	<ul> <li>Solvents</li> </ul>	